

Buffer solution – pH working standard of 2nd category

INTENDED FOR

reproducing and transmitting the activity index of hydrogen ions (pH) in aqueous solutions; verification and calibration of pH measuring instruments; controlling the metrological characteristics when tested, and for type approval; error control of measurement procedures (MP) pH of liquid media.



Buffer solutions are aqueous solutions of chemicals that are packaged in plastic vials. The bottle is sealed and has a screw

cap wit a protective ring to prevent unauthorized access. Buffer solutions have 9 modifications, that differ by concentration of chemicals and reproducible values.

Limits of permissible absolute accuracy of pH reproduction:	
in the temperature range from + 5 to + 20°C	±0.02
in the temperature range from + 20 to + 40°C	±0.01
in the temperature range from + 40 to + 80°C	±0.02
Bottle volume	125 cm ³ , 270 cm ³
Expiry date	1 year



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Marking of buffer solu- tions	Marking of buffer solutions	Chemical formula of substances	Substance oncentration	pH reproducible value at temperature +25°C
BS-pH-1	Potassium tetraoxalate dihydrate	KH ₃ (C ₂ O ₄) ₂ ·2H ₂ O	0.1	1.48
BS-pH-2	Potassium tetraoxalate dihydrate	KH ₃ (C ₂ O ₄) ₂ ·2H ₂ O	0.05	1.65
BS-pH-3	Potassium hydrogenphthalate	KHC ₈ H ₄ O ₄	0.05	4.01
BS-pH-4	Potassium phosphate monosubstituted Sodium phosphate twice-substituted	KH ₂ PO ₄ Na ₂ HPO ₄	0.025 0.025	6.86
BS-pH-5	Potassium phosphate monosubstituted Sodium phosphate twice-substituted	KH ₂ PO ₄ Na ₂ HPO ₄	0.0087 0.0304	7.41
BS-pH-6	Potassium phosphate monosubstituted Sodium phosphate twice-substituted	Na ₂ B ₄ O ₇ °10H ₂ O	0.01 0.04	7.43
BS-pH-7	Sodium tetraborate decahydrate	Na ₂ B ₄ O ₇ °10H ₂ O	0.01	9.18
BS-pH-8	Sodium tetraborate decahydrate	Na ₂ B ₄ O ₇ °10H ₂ O	0.05	9.18
BS-pH-9	Sodium carbonate Sodium carbonate acid	Na ₂ CO ₃ NaHCO ₃	0.025 0.025	10.00



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